# K. N. Bhise Arts and Commerce College, Kurduwadi

## **Department of Chemistry**

#### **Programme Outcomes**

Sr.	Programme	Programme Objectives	Programme Specific Outcomes
No.			
1	B. Sc.	PO1. Scientific attitude:	<b>PSO1.</b> Understand the knowledge of
	Chemistry	To develop our thinking, our	chemistry, Chemistry majors are able to
		attitude, our interests, our outlook	recognize and apply the principles of
		have undergone tremendous	atomic and molecular structure to predict
		changes.	chemical properties and chemical
			reactivity.
		<b>PO2. Effective Communication:</b>	<b>PSO2.</b> Chemistry majors are able to
		Speak, read, write and listen	employ critical thinking and scientific
		clearly in person through	enquiry in the performance, design,
		electronic media in English and in	interpretation and documentation of
		one Indian language and make	laboratory experiments, at a level suitable
		meaning of the word by	to succeed at an entry level position in a
		media and technology	programme
		PO3 Laboratory and calculation	<b>PSO3</b> Upon completion of a chemistry
		skills. Demonstrate competency in	degree chemistry majors are able to
		the laboratory expected of a	interpret and analyze quantitative data
		practicing chemist. Develop good	
		skills in laboratory such as	
		observation and evaluation.	
		PO4. Integrate the big ideas and	<b>PSO4.</b> The students are able to understand
		themes:	theoretical concepts of instruments that are
		Structure representation,	commonly used in most chemistry fields as
		energetics, rates of processes,	well as interpret and use data generated in
		Connect neutral network across	instrumental chemical analyses.
		courses.	
		PO5. Effective use of ICT:	<b>PSO5.</b> Application of chemical knowledge
		Use of ICT tools to communicate,	to explain observation and solve the
		create, disseminate, store and	problem.
		manage information.	

# K. N. Bhise Arts and Commerce College, Kurduwadi

## **Department of Chemistry**

#### **Course Outcomes**

Sr.	C	ourse	Course Outcomes
1.	B. Sc I	<b>Paper-I</b> Physical Chemistry	<ol> <li>To study the basic concepts order and molecularity of reaction.</li> <li>To understand the methods to determine the order of reaction.</li> <li>To study the basic concepts of graphical representation, Derivative and Integration.</li> <li>To introduce the processes, laws of thermodynamics.</li> <li>Explanation of ideal, non ideal and liquification of gases.</li> </ol>
	Sem- I	<b>Paper-II</b> Inorganic Chemistry	<ol> <li>To understand the atomic structure, shape of orbital, position of elements in periodic table, electronic configuration and characteristics.</li> <li>Introduction of chemical bonding, types of bond and crystal structure of NaCl &amp; CsCl, Unit cell.</li> <li>To study the VBT, understand concept of hybridization.</li> <li>To study the MOT, understand the concept of orbital's.</li> <li>VSEPR Theory- assumption and application.</li> </ol>
2.	B. Sc I	<b>Paper-III</b> Organic Chemistry	<ol> <li>Fundamentals of organic chemistry- bond energy, bond angle, bond length, types of arrow, types of reactions.</li> <li>To understand the Structure and Bonding in organic compounds.</li> <li>To introduce the term isomerism, nomenclature.</li> <li>To identify Aromatic, Non-aromatic and Anti-aromatic compounds by using Huckels rule.</li> <li>To study the alkenes, dienes and alkynes.</li> </ol>
	Sem- II	<b>Paper-IV</b> Analytical Chemistry	<ol> <li>To introduce physical properties of liquids.</li> <li>Introduction of environmental chemistry and different treatments on water.</li> <li>To study qualitative and quantitative analysis. detection of Carbon, Hydrogen, Nitrogen And Halogen in different organic compounds by qualitative and quantitative measurements.</li> <li>To study the petrochemicalcomposition of petrochemicals, application of petrochemicals.</li> </ol>
3.	B. Sc I	Practical	<ol> <li>To introduce the basic concepts of Physical, Inorganic and Organic chemistry practicals.</li> <li>To provide practical information to students about titration, preparation and estimations.</li> <li>To expose students to handle the glassware, instruments and apparatus.</li> <li>To improve the practical knowledge of the students</li> </ol>
4.	B. Sc II Sem- III	<b>Paper-V</b> Organic Chemistry	<ol> <li>To introduce the UV-Visible spectroscopy.</li> <li>To study the conformation, configuration and isomerism.</li> <li>Brief idea about name reactions with mechanism.</li> <li>To study of alcohols, phenols, aldehydes, ketones, ethers, epoxides and carboxylic acids.</li> <li>To introduce various dye i.e. Methyl orange, Congo-red.</li> <li>To study the co-ordination chemistry-to understand the</li> </ol>

			Ligand, co-ordination number, Werners theory.
			2. To introduce chelation, structure of DMG and EDTA.
		Paper-VI	3. To introduce acid & bases and different concepts.
		Inorganic	4 To study of d-block elements- position electronic
		Chemistry	configuration and characteristics of d block elements
		Chemistry	1. The stellar of Desistence on 1 Can between
			1. The study of Resistance and Conductance.
			2. Knowledge of transport number, kohlrauschs law and its
		Paper-VII	applications, Onsagar equation.
		Physical	3. Describe the term Entropy, second law and third law of
		Chemistry	thermodynamics.
			4. To study the cell, lattices, crystal structure.
5.	B. Sc II		5. Introduction of distribution law.
	Sem- IV		1 To introduce volumetric analysis-understand term of
			titrant titrand mole concent primary besecond standard
		Danar VIII	whateness To study said hass indicatene basid hass theory
		raper-viii	substance. To study actu base indicators & actu base theory.
		Analytical	2.10 introduce gravimetic analysis-understand the general
		Chemistry	steps dilution, precipitation, digestion, ignition in
			gravimetric analysis. To understand the co-precipitation &
			post precipitation.
			1. To synthesis, estimation and analysis of unknown
			organic compounds.
			2 To prepare and estimate gravimetrically inorganic
6	R Se II	Practical	compounds
0.	D. 5011	Tacucai	2. To detact the universe soldie and basic redicals
			4. To useful the unknown actual and basic faultais.
			4. To perform the experiments by instrumentally and non-
			instrumentally.
			5. To study conductivity, refractivity, viscosity, specific
			rotation practically.
			1. To understand Phase rule, phase diagram, one
			component and two component systems.
		Paper-IX	2.To acquire a knowledge of Thermodynamics, types of
		Physical	electrodes metal- metal ion electrode
		Chemistry	3 explain the Ovidation-Reduction electrodes reversible
		Chemistry	and irreversible cell and applications of emf
_			4. Explain the thermal and photochemical processes, Laws
7.	B. Sc III		of photochemistry and Joblonski diagram.
	Sem- V		1.To understand the basic concept, factors, applications
			and limitations of CFT.
			2. Describe the basic concepts, assumptions, examples,
		Paper-X	applications and limitations of MOT.
		Inorganic	3. To understand the nuclear reaction and its types.
		Chemistry	annlications radioisotones
		Chemistry	4 To study the essential non essential and trace elements
			in biological process
			The description of the last second se
			5. To describe the basic concepts of polymers, and
			nanoparticles.
			1. Brief idea about Spectroscopy i.e. IR, NMR, Mass.
			2.To introduce Shielding and Deshielding, Chemical Shift.
		Paper-XI	3. To describe various conformations- chair, boat, half-
		Organic	chair, twist-boat etc.
		Chemistry	4. To understand the name reactions with mechanism
			5. To synthesis and reactions of AAF and Malonic ester
			1 To provide basic information of colorimetry
			2. to growing basic information of colorintetry.
1	1	1	2.10 study the thranons by potentionnetrically.

Paper-XII	3. Explain the electrolysis and electroplating.
Analytical	4. Discuss the Flame photometry and titrations by
Chemistry	conductometrically.

		<b>Paper-XIII</b> Physical Chemistry	<ol> <li>To introduce the Rotational and Vibrational spectroscopy of diatomic molecules.</li> <li>To study the basic ideas about different solutions.</li> <li>Explain the terms- free energy, enthalpy, phase equilibria, law of mass action, fugacity and activity.</li> <li>Describe effects on rate of reactions, theories and order of reactions.</li> </ol>
8.	B. Sc III Sem- VI	<b>Paper-XIV</b> Inorganic Chemistry	<ol> <li>To study the Lanthanides and Actinides with IUPAC nomenclature of super heavy elements.</li> <li>To understand the theories of bonding in metal, Types of Semiconductor, applications of Semiconductor and Superconductor.</li> <li>Recognise the structural studies of Boranes, Xenon compounds and oxides of sulphur and Phosphorus.</li> <li>To study the corrosion of metals and Passivity.</li> <li>To introduce Organomettalic chemistry.</li> </ol>
		<b>Paper-XV</b> Organic Chemistry	<ol> <li>To study the Five membered, Six membered and Condensed heterocycles.</li> <li>Brief idea about the monosaccharide, disaccharide and polysaccharide.</li> <li>To understand the concept Dyes of Drugs.</li> <li>To understand the analytical evidences and synthesis of Vitamins and Hormones.</li> <li>To study the pesticides and plant growth.</li> </ol>
		<b>Paper-XVI</b> Analytical Chemistry	<ol> <li>To introduce soap and detergents.</li> <li>Describe the process of polymerisation and rubber.</li> <li>To explain the manufacture, refining of sugars and their by- products.</li> <li>To study the Sizing, Bleaching and Dyeing of fibres.</li> <li>To study the basic of Green Chemistry and Chromatography.</li> </ol>
9.	B. Sc III	Practical	<ol> <li>Apply quantitative reasoning skills to determine quantities of matter and energy involved in physical and chemical change.</li> <li>Analyse and interpret experimental data using instruments.</li> <li>To synthesize and Estimate Inorganic compounds.</li> <li>To synthesize and estimate Organic compounds.</li> <li>Separation and analysis of binary organic mixtures.</li> <li>Preparation of derivatives of Organic compounds.</li> </ol>